

Production and Stability Assessment of Oxygen Carrier Produced By Sewage Sludge Fluidized Bed Combustion

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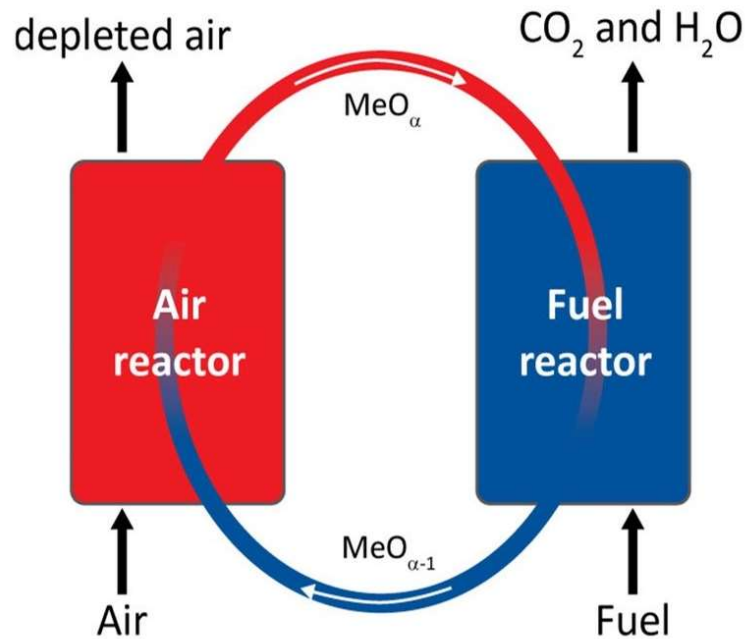
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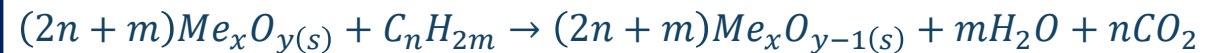
Introduction: Chemical Looping Combustion



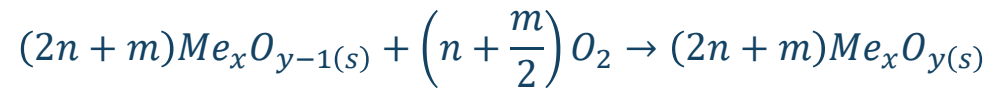
The Chemical Looping Combustion is an innovative technology that allows to burn fuel avoiding the direct contact between it and the oxygen of the air. The process is based on the use of an Oxygen Carrier (OC, typically a metal oxide) to transfer oxygen from the air to the fuel, allowing the formation of burnt gases free of nitrogen and therefore composed mainly of carbon dioxide (concentrated) and water (separable).



Fuel Reactor



Air Reactor



Properties required of oxygen carriers:

- High reactivity
- High oxygen transport
- Low costs
- Low environmental impact
- High mechanical strength

Introduction: Oxygen Carrier



Typical values of R_o , oxygen ratio (oxygen transport capacity)

- Fe-based OC, $R_o=0.045$
- Cu-based OC, $R_o=0.089$
- Ni-based OC, $R_o=0.091$

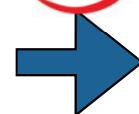
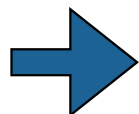
$$R_o = \frac{m_o - m_r}{m_o}$$

What are the limits?

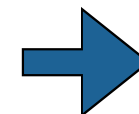
- High carrier cost
- Carrier preparation method



Test a low-cost OC by re-using sewage sludge, and evaluate performance in a Chemical Looping Combustion process. The OC was prepared by combustion of sludge in a fluidized bed, and deposition of the ashes on a support of γ -Alumina (Al_2O_3).



**OXYGEN
CARRIER**



**CHEMICAL
LOOPING
COMBUSTION**

Aim of the work



- The metals (Fe, Ca, Mn) present in sewage sludge ash have been deposited during sewage sludge fluidized bed combustion on γ -alumina which is characterized by high mechanical resistance and it was used as bed material.
- Three different combustion tests have been carried out in a 41 mm ID reactor adopting different operative conditions: the fluidization velocity and the particle size, in order to find optimal conditions for the preparation of the OCs.
- The oxygen carriers has been tested in reduction/oxidation tests using methane as fuel.
- The performance of the prepared OCs have been performed using a lab-scale fluidized bed apparatus properly designed for the study of chemical looping processes as close as possible to reality, in terms of cycling of temperatures and of reaction atmospheres.
- The samples have been characterized by TPR analysis, carrying out reduction tests with a mixture of 2% H₂/Ar, and a temperature ramp of 10°C/min up to 850°C.

Materials



Chemical composition

Proximate analysis		Elementar analysis (dry basis)	
Moisture	12,2	Carbon	39,4
Fixed carbon	57,4	Hydrogen	5,6
Volatiles	9,3	Nitrogen	6,7
Ash	21,2	Sulfur	0,8
		Chlorine	0,7
		Oxygen	22,7
		Ash	24,1
Higher heating value, kJ/kg		13484	
Lower heating value, kJ/kg		12135	

Inorganic elemental analysis of sewage sludge (as received), mg/kg

	%		ppm		ppm
Al	0,694	Ba	130	Ni	68
Ca	1,378	B	55	Se	3
Fe	2,671	Cr	81	Sr	70
Mg	0,360	Co	4	Sn	18
P	1,496	Cu	294	Ti	202
K	0,202	Pb	69	Tl	0,03
Si	0,255	Mn	247	V	15
Na	0,043	Hg	3	Zr	3
Zn	0,059	Mo	6		

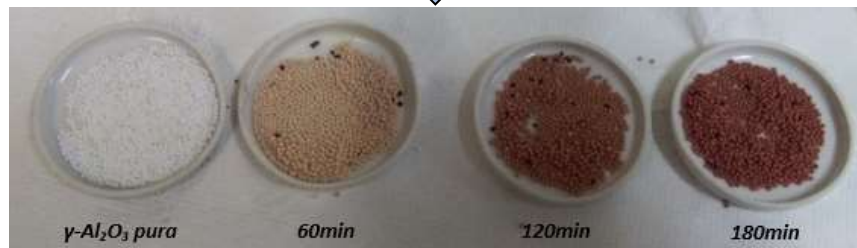
Experimental Combustion tests



The experiment was carried out in a bubbling fluidised bed reactor in laboratory scale employing air as fluidization gas, using the sludge as fuel and alumina as an inert material bed.

Operative conditions	Test 1	Test 2	Test 3
Temperature [°C]	850	850	850
U_{gas} [m/s]	0.47	0.6	0.6
$\langle d \rangle$ alumina bed [μm]	1000	1000	400–600
Feed [g/h]	100	120	120
Bed weight [g]	180	180	180

Each test was approximately carried out for 3 hours. Three samples were taken, in order to evaluate the effect of time on the degree of covering sludge / alumina ashes

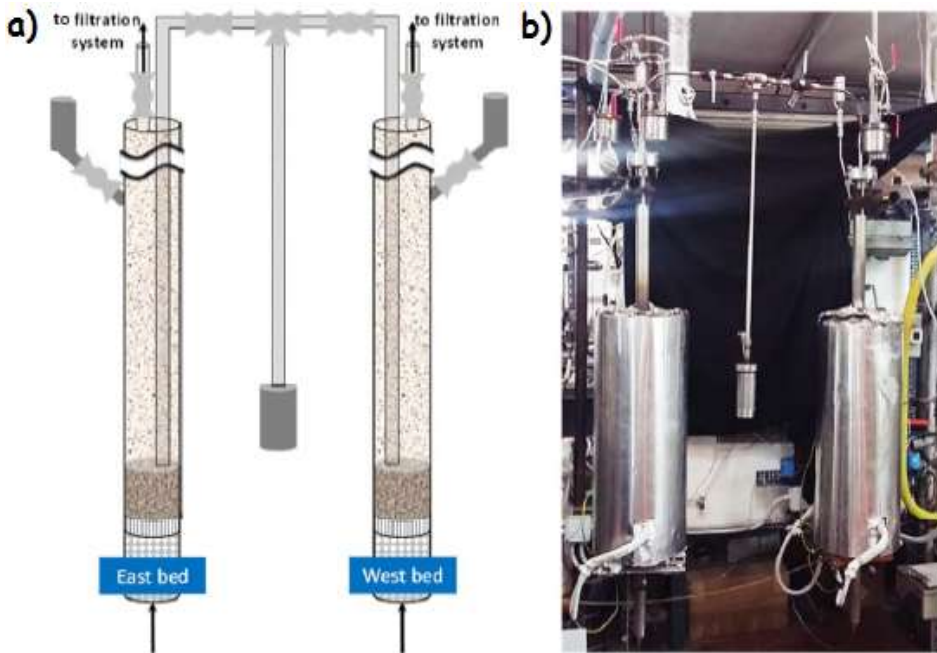


Fluidized bed reactor

Experimental :CLC tests



TWIN BEDS Apparatus



	Air Reactor	Fuel Reactor
$T, ^\circ\text{C}$	850	750/800/850
$Q, \text{NL/h}$	617	677/646/617
$U, \text{m/s}$	0.5	
Fluidization Gas	Air	2%vol CH_4/N_2
Reduction/Oxidation cycles	5	
Time, min	10	25

Experimental: TPR analysis and reduction/oxidation cycles



Micromeritics Autochem II TPD/TPR analyzer

110 mg OC

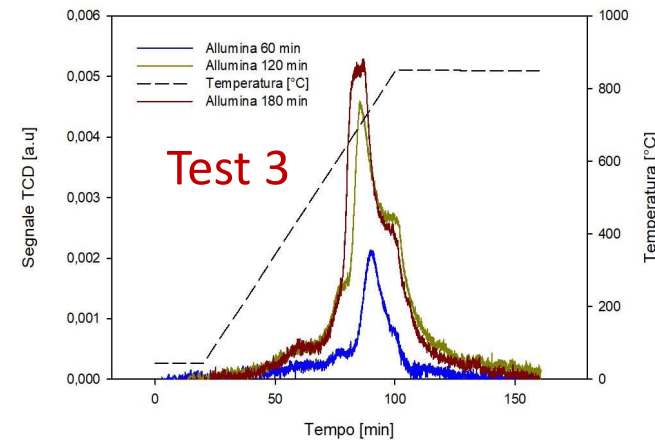
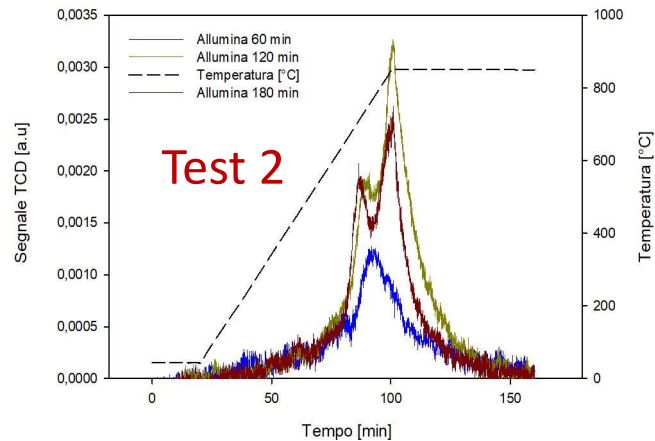
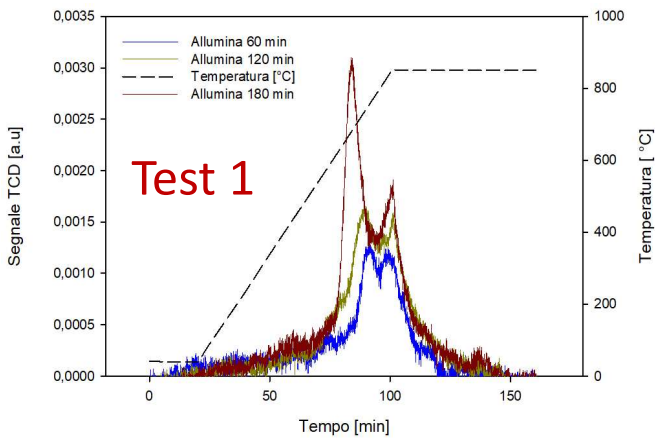
2% H₂/Ar mixture (50 Ncm³ min⁻¹)

10 °C min⁻¹ up to **850 °C**.

1 h in air (50 Ncm³ min⁻¹) at 850 °C before the experiment.

The samples have been characterized by TPR (temperature programmed reduction) analysis, to study the reactivity of the carriers and to evaluate the amount of reducible material deposited on the support.

Results Combustion tests

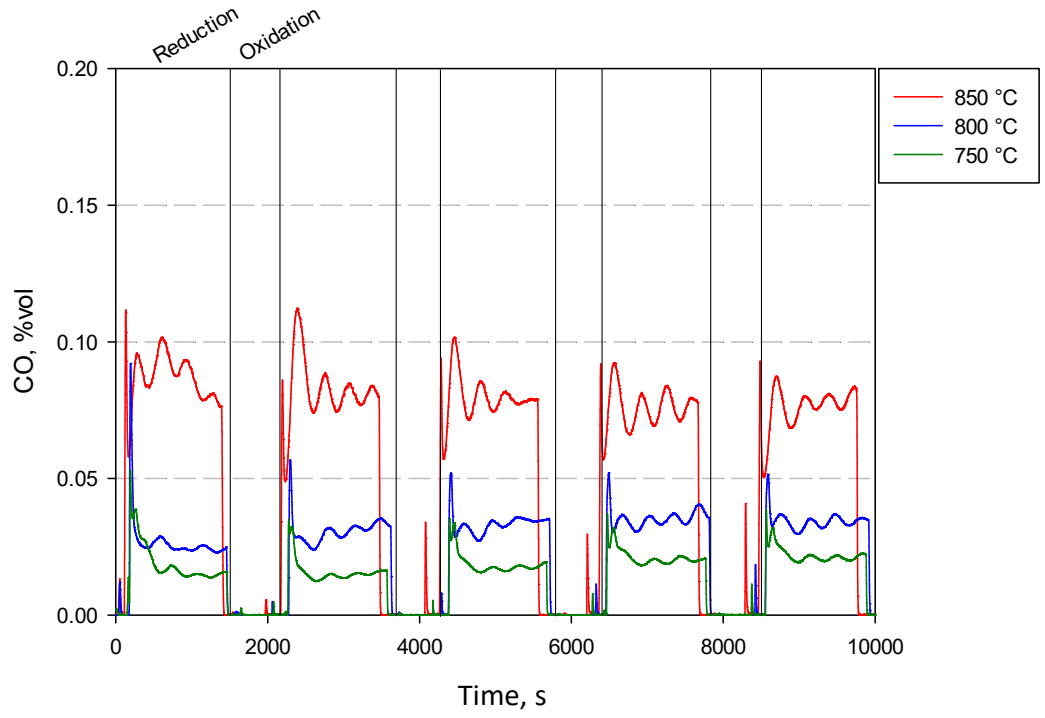
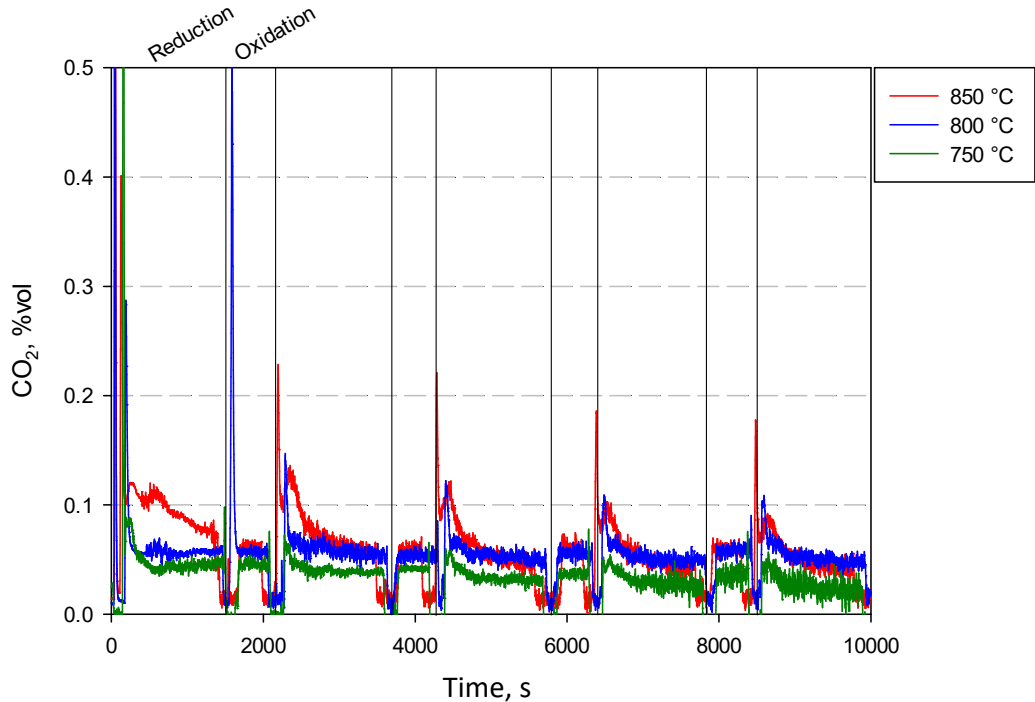
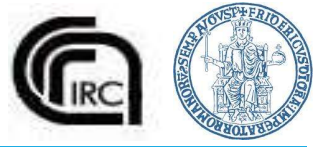


	60 min		120 min		180 min	
	μmol H ₂	mg H ₂ /g _c	μmol H ₂	mg H ₂ /g _c	μmol H ₂	mg H ₂ /g _c
Test 1	7.26	0.13	10.17	0.19	13.46	0.26
Test 2	8.67	0.16	16.41	0.30	12.88	0.24
Test 3	7.47	0.13	23.61	0.38	24.11	0.43

$$R_o = 3.4 \times 10^{-3}$$

Results

CLC tests



The reactivity of the OC increases with temperature and decreases as the number of cycles grows.

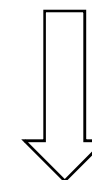


Results

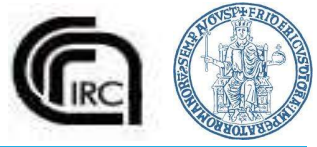
CLC tests

<i>T</i> Fuel Reactor	Cycle	CH ₄ conversion degree, %	mmol O/ g OC	mmol CO/ g OC	mmol CO ₂ / g OC
850°C	1	10.0	1.582	0.237	0.218
	2	8.3	1.282	0.226	0.151
	3	8.3	1.163	0.217	0.128
	4	8.8	1.040	0.202	0.109
	5	10.1	1.107	0.198	0.128
800°C	1	10.4	0.713	0.071	0.125
	2	5.9	0.596	0.085	0.085
	3	8.6	0.753	0.090	0.121
	4	7.4	0.746	0.095	0.115
	5	8.7	0.785	0.092	0.127
750°C	1	3.8	0.285	0.045	0.037
	2	4.1	0.654	0.048	0.127
	3	4.0	0.603	0.057	0.108
	4	3.9	0.581	0.065	0.096
	5	5.0	0.531	0.067	0.083

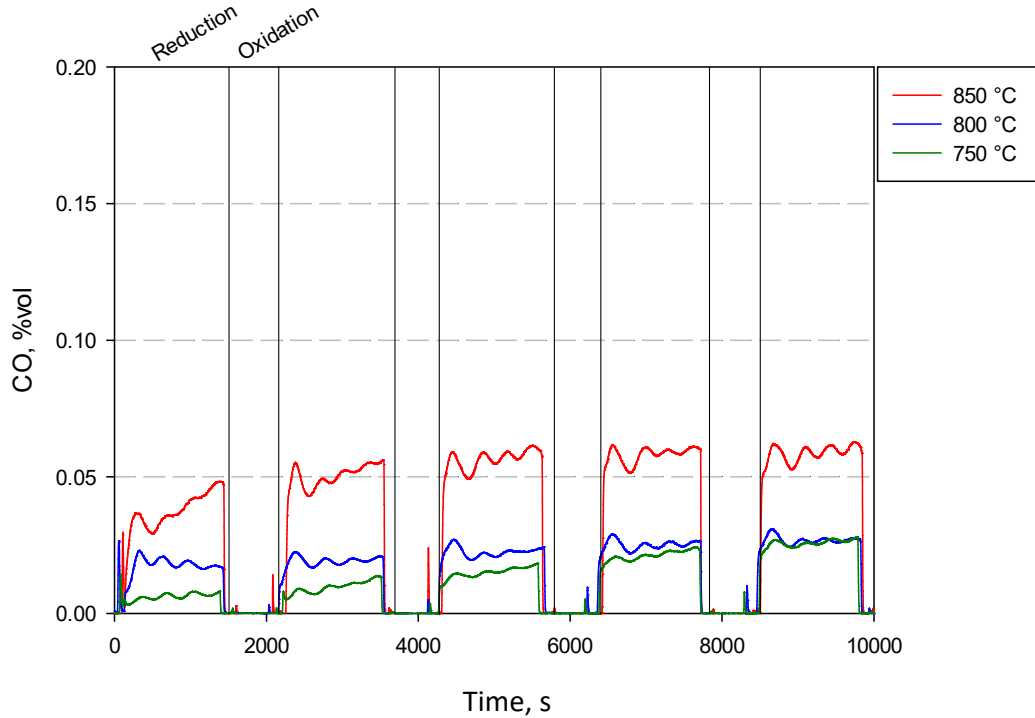
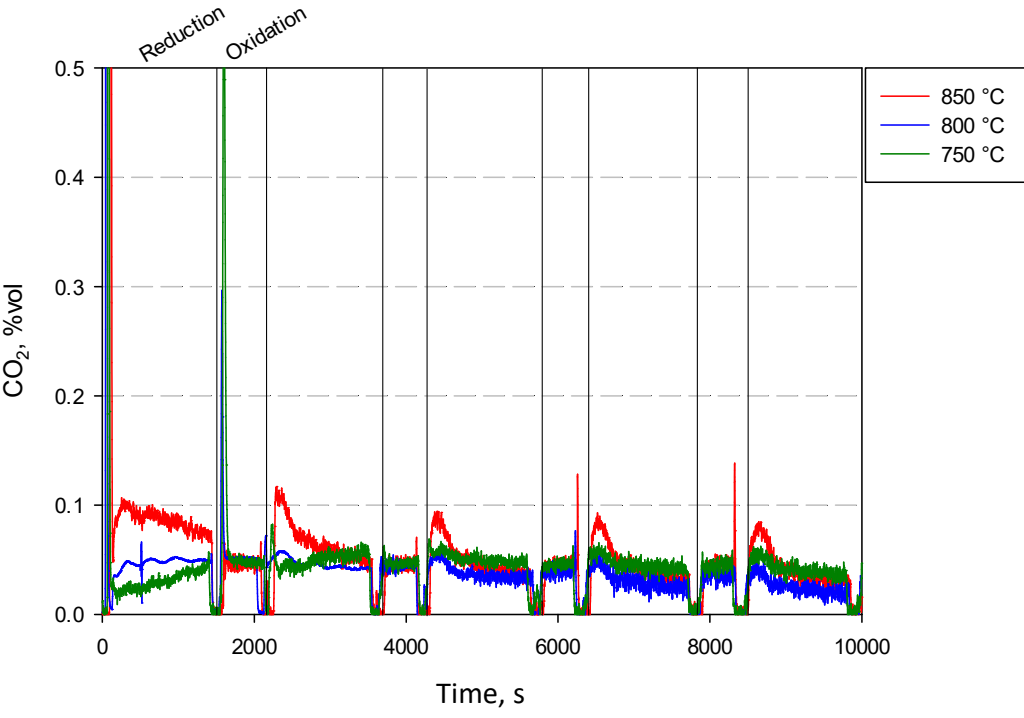
- $R_{O,max} = 25.3 \times 10^{-3}$
- Fresh material:
 4.5×10^{-3}



Support reactivity?



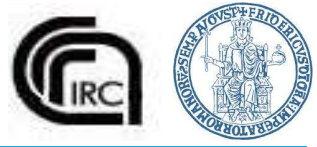
Results (using the support only)



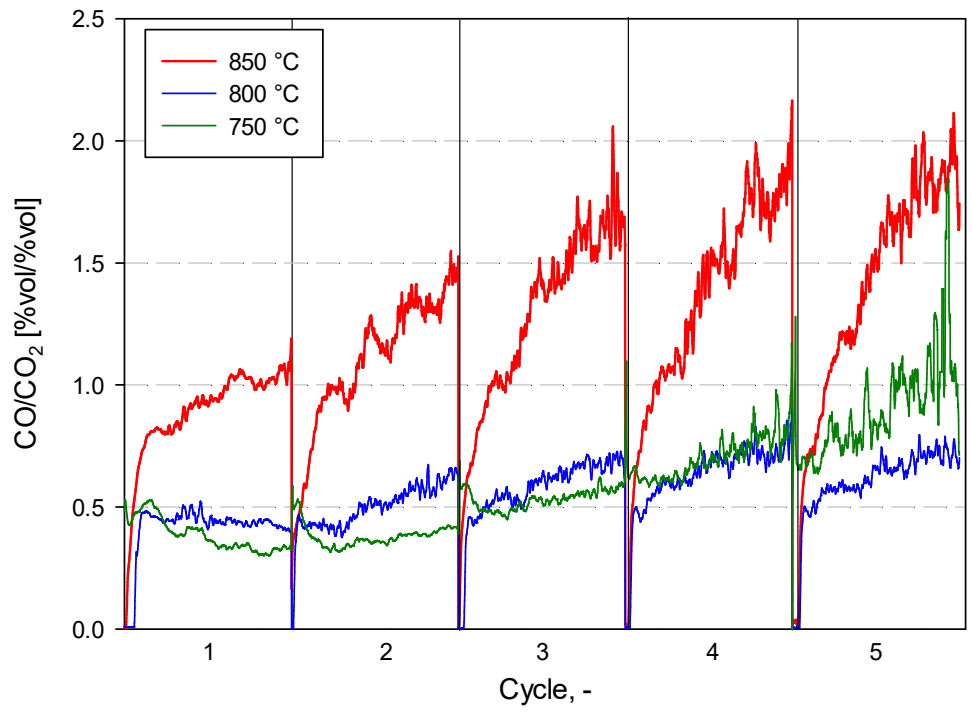
Alumina reactivity increases with the temperature increase and with increasing the number of cycles.

Results

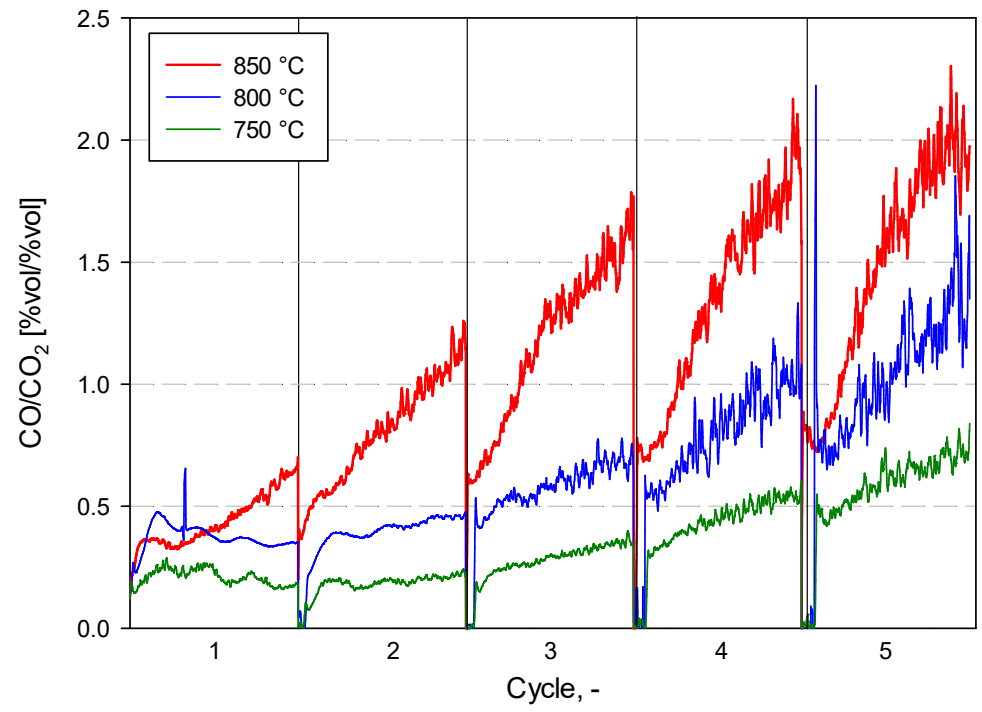
:CLC tests



Oxygen Carrier

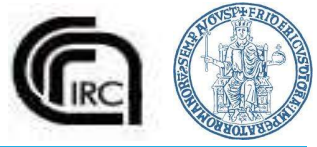


Allumina

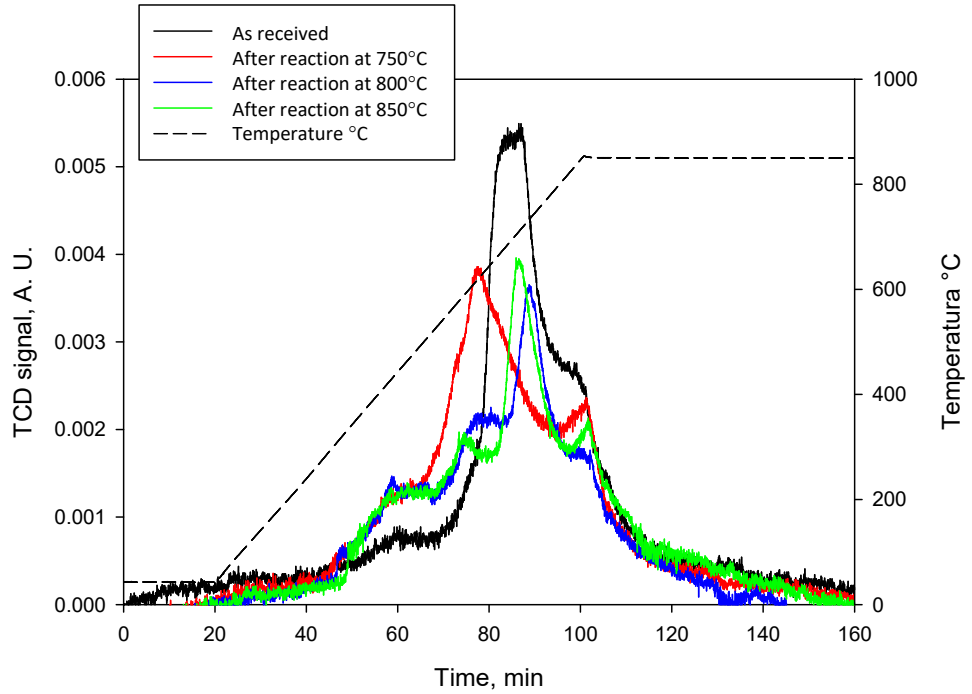


Results

:CLC tests

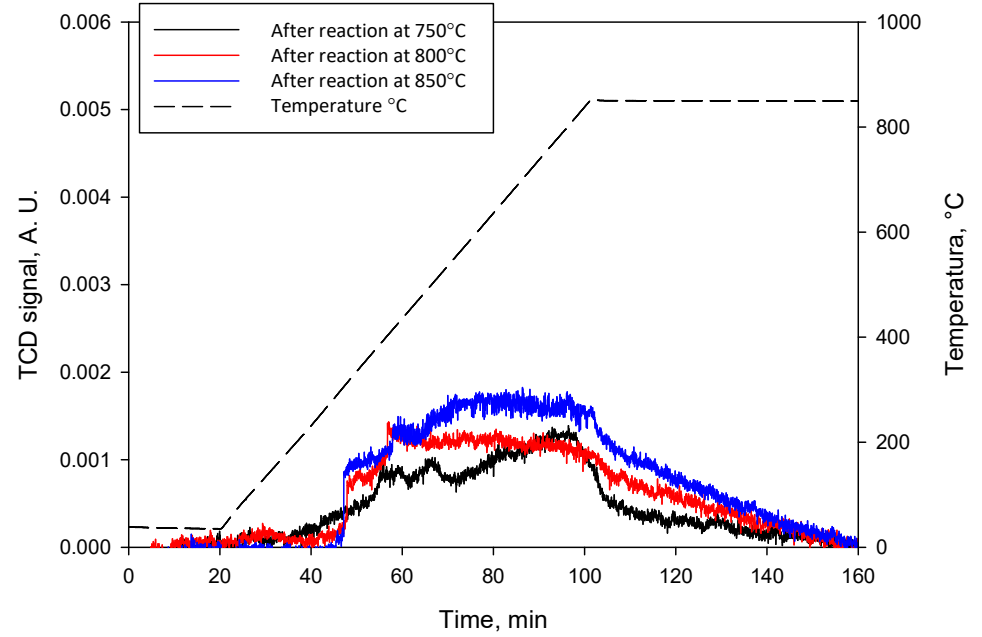


Oxygen Carrier



The material undergoes structural changes

Alumina



Unchanged profile

Conclusions



Sewage sludge fluidized bed combustion can be considered a feasible and reliable strategy for the preparation of low-cost CLC oxygen carriers.

The reactivity of the OC increases with temperature increase and decreases with the number of cycles

$R_{O,max}=25.4 \times 10^{-3}$, lower than the values reported in the literature → it is necessary to improve the deposition process.

From the TPR analysis we note that the material undergoes structural changes following the looping process.



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***Thank you for your
kind attention***