



# **Towards Sustainable Energy : Carbon Capture, Utilization and Storage**

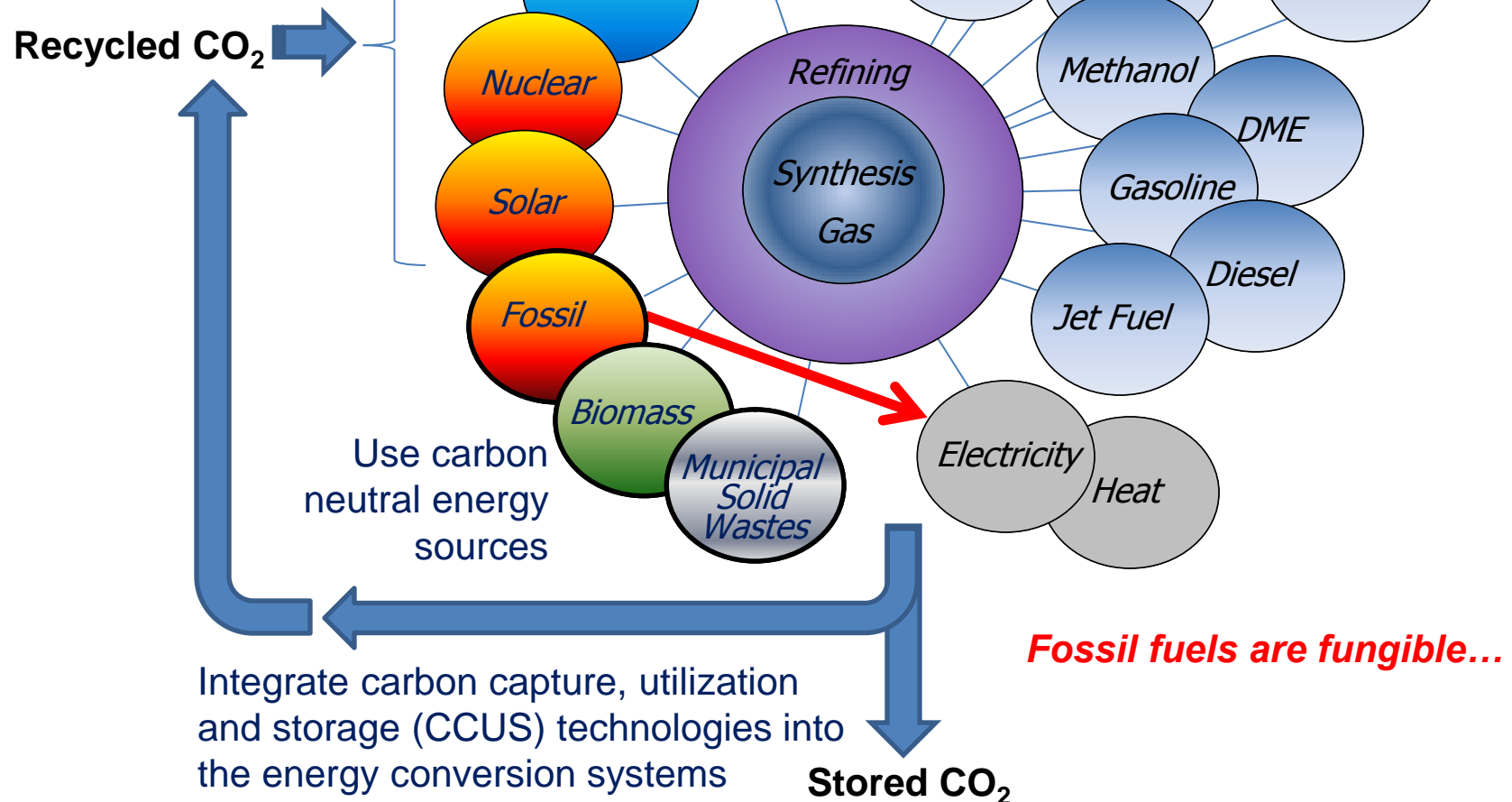
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Lenfest Center for Sustainable Energy  
Columbia University

NSF RCN-CCUS Annual Meeting  
April 16<sup>th</sup>, 2014

# Towards Sustainable Energy and Environment

Use domestic energy sources to achieve energy independence with environmental sustainability



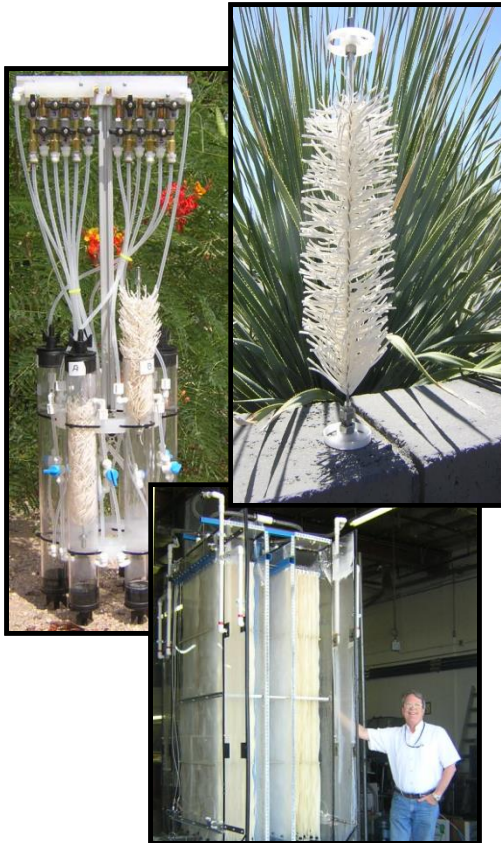
# Carbon Capture

- From diffuse sources
- From concentrated sources
  - Physical and chemical absorption and adsorption, Cryogenic separation, membrane separation, reaction-based sorbent injection
  - Oxyfuel combustion
  - **Integrated Carbon Capture Technologies:** ZECA, HyPr-Ring process, ALSTOM process, GE fuel-flexible process, Calcium looping process, Coal-direct chemical looping reforming process and Syngas redox process, Membrane process, etc

# Collecting CO<sub>2</sub> with Synthetic Trees

## From Technology Validation to Market-Flexible Products to Scalable Global Solutions

### GRT Pre-Prototype Air Capture Modules - 2007



### Current GRT Development



### Mass-Manufactured Air Capture Units



Courtesy GRT\*

\*K. S. Lackner is a member of GRT

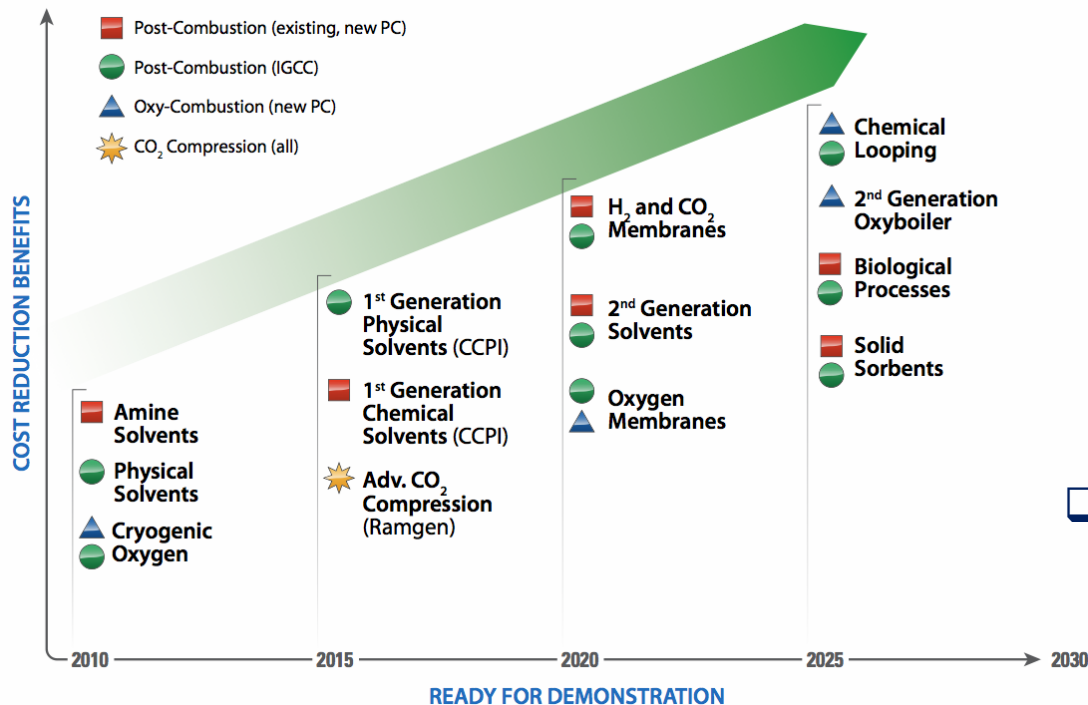
# Carbon Capture, Utilization and Storage Technologies (CCUS)

Utilization

Capture

Storage

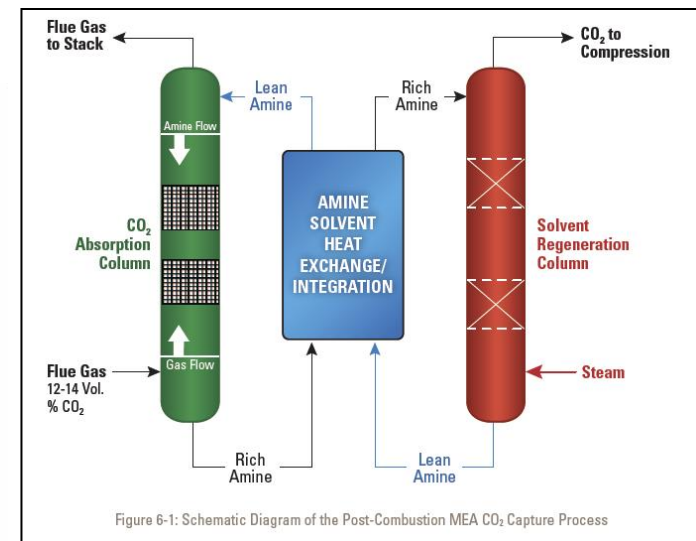
## Carbon Capture Technologies



(NETL, 2010)

## Required characteristics for CCS

- Capacity and economic feasibility
- Environmental benign fate
- Long term stability



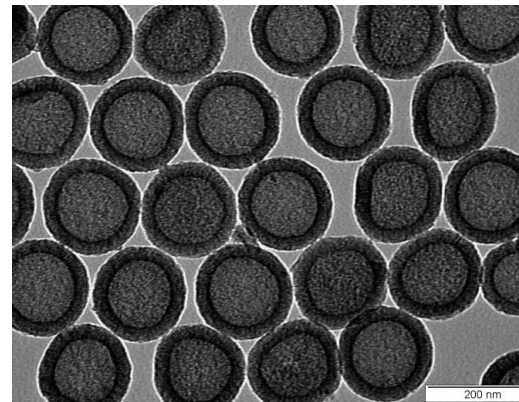
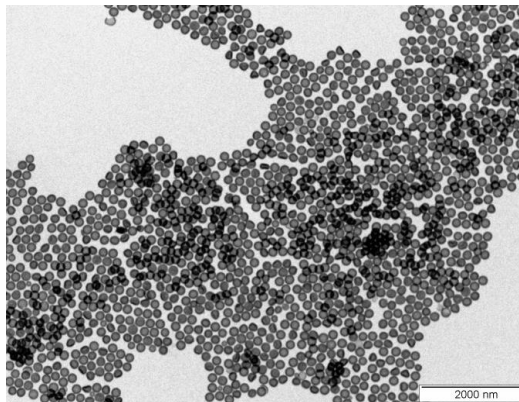
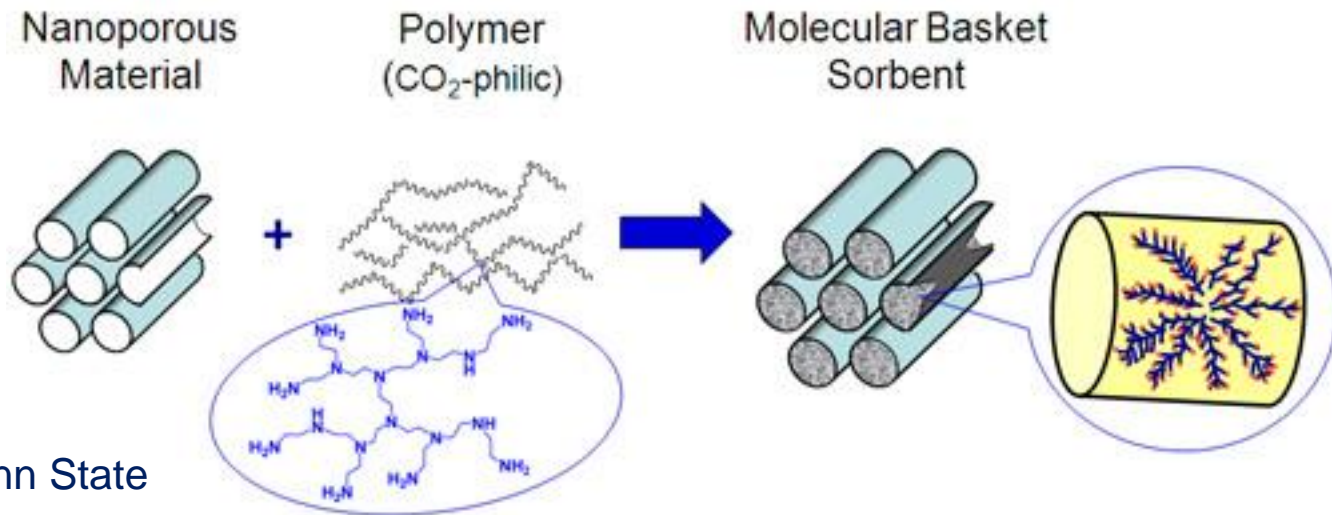
(NETL, 2011)

## MEA Challenges

- Corrosion and solvent degradation
- High capital and operating costs
- High parasitic energy penalty

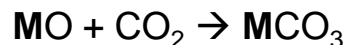
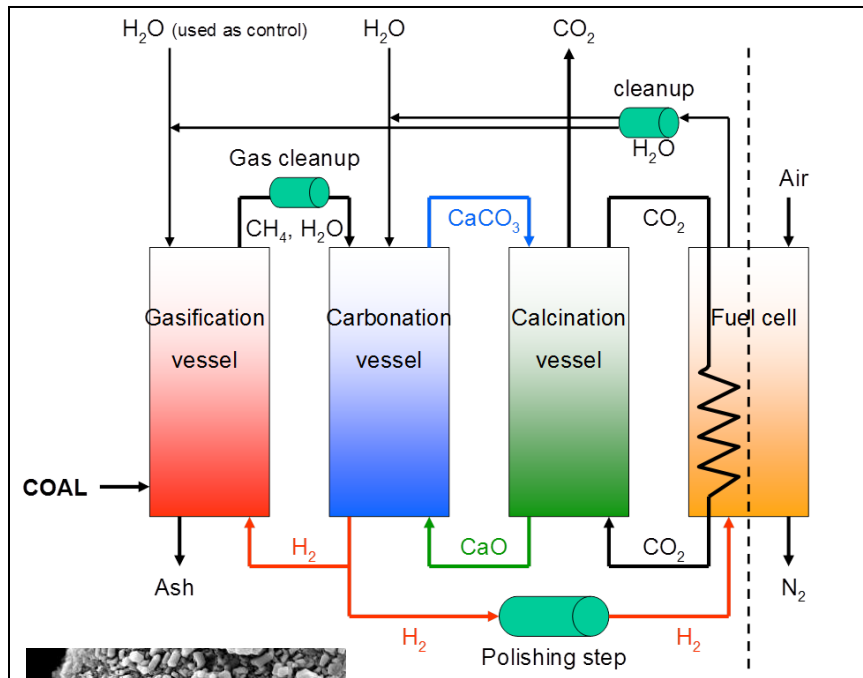


# Novel CO<sub>2</sub> Capture Materials



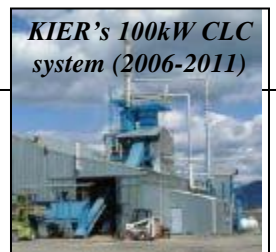
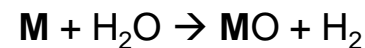
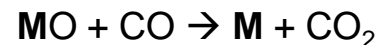
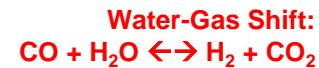
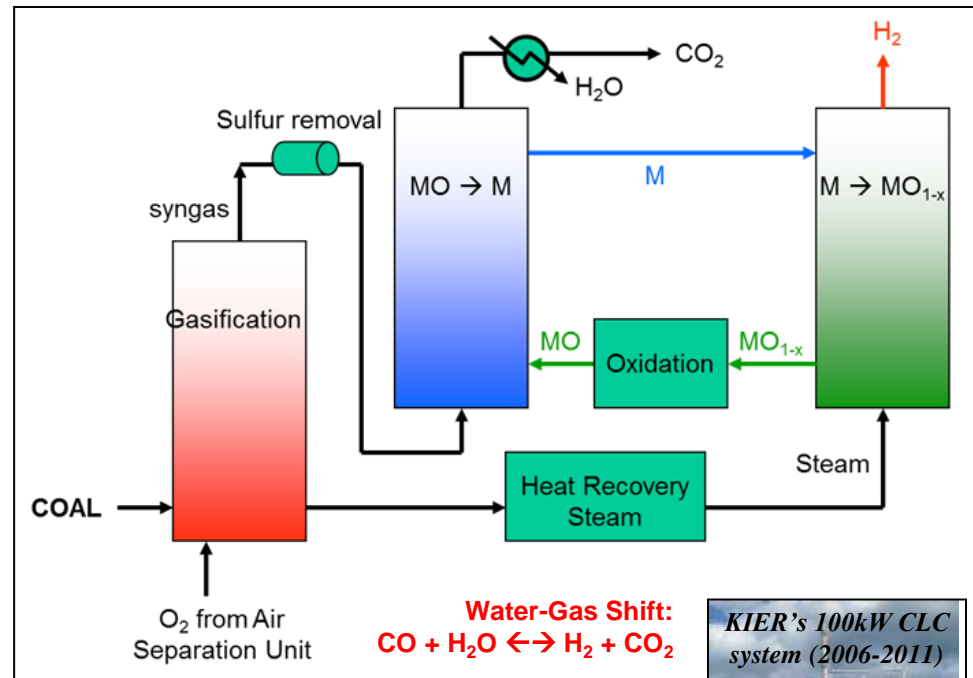
# Solid Sorbents & Chemical Looping Technologies

## Carbonation / Calcination cycle



e.g., ZECA process  
(Los Alamos National Lab)

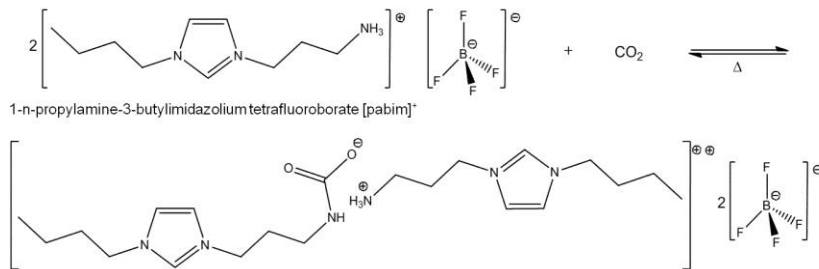
## Oxidation / Reduction cycle



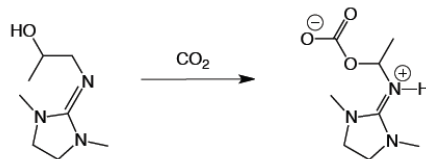
e.g., Chemical Looping process for H<sub>2</sub> production  
(Ohio State Univ.: U.S. Patent No. 11/010,648 (2004))

# Novel CO<sub>2</sub> Capture Solvents (NETL and ARPA-E funded projects)

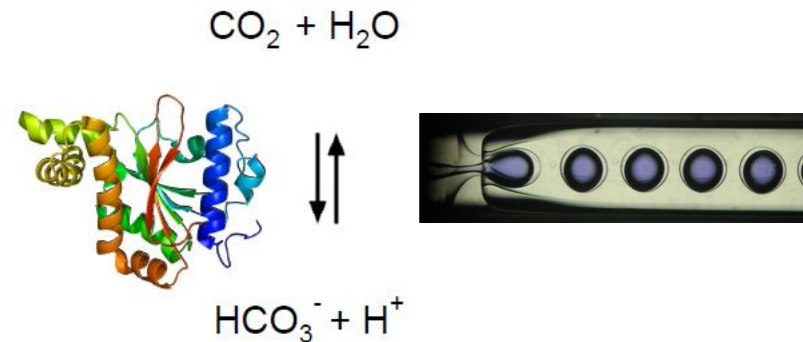
## □ Ionic liquids



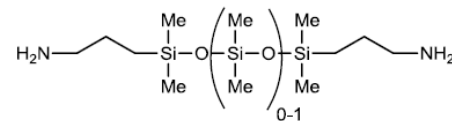
## □ CO<sub>2</sub>BOLs



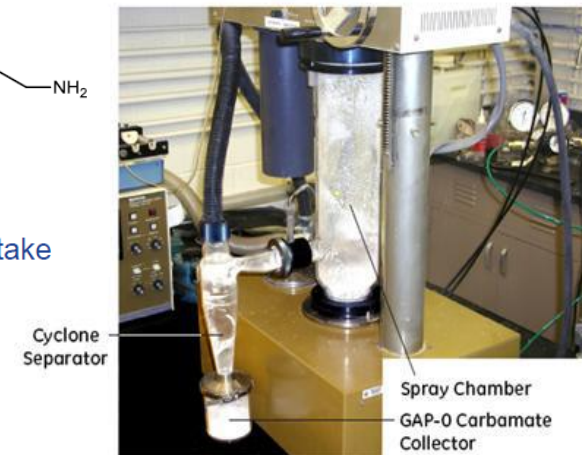
## □ Carbonic Anhydrase (Enzyme)



## □ Phase changing absorbents



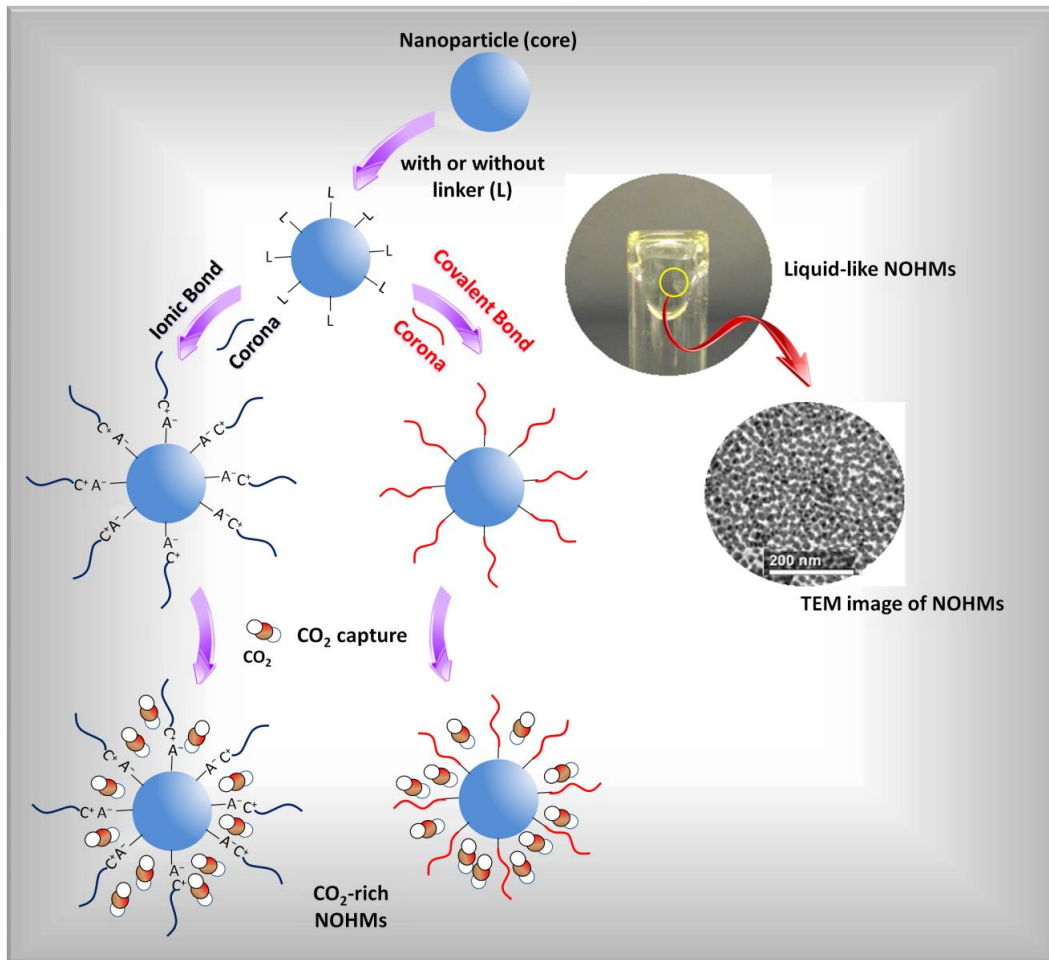
GAP-0/1  
13.1 - 17.3% CO<sub>2</sub> uptake



## □ Liquid-like Nanoparticle Organic Hybrid Materials



# Nanoparticle Organic Hybrid Materials (NOHMs)



## ☐ Solvent-free liquid-like hybrid systems

- Solvent tethered to nanoparticle cores
- Zero-vapor pressure and improved thermal stability

## ☐ Tunable chemical and physical properties

- Liquid, solid, gel
- Solvation in NOHMs driven by both entropic and enthalpic interactions

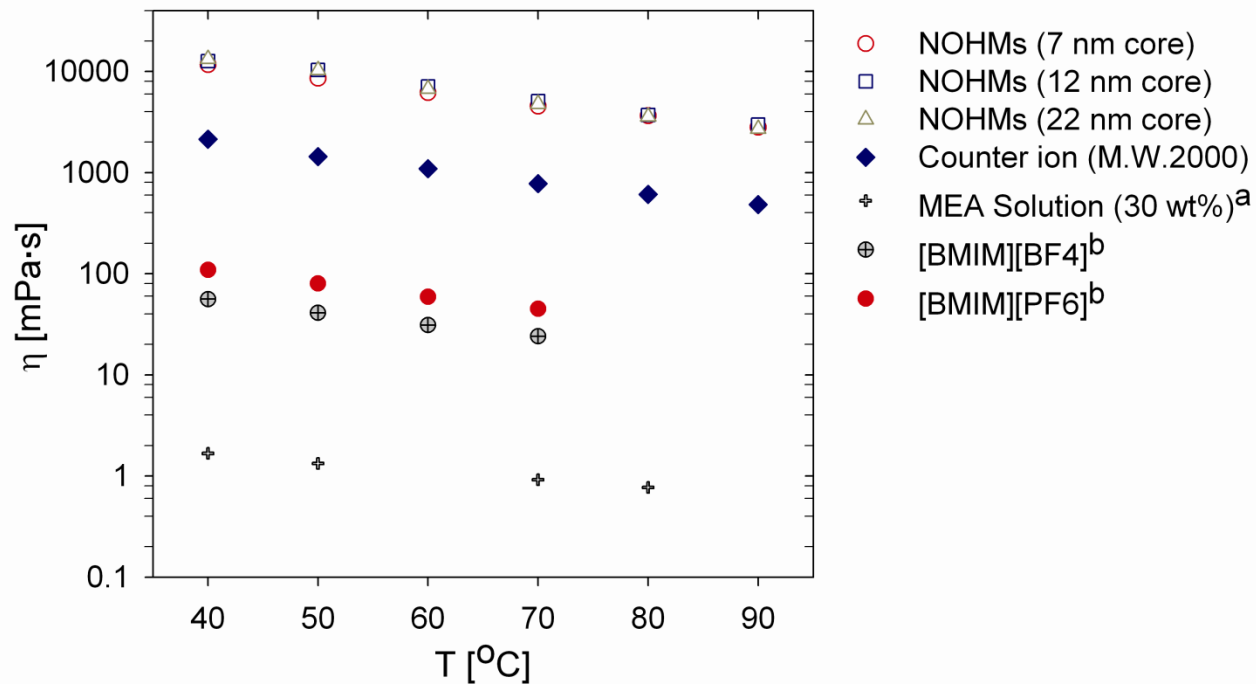
## ☐ Straightforward synthesis

- Easy to scale up

**ZERO VAPOR PRESSURE+ TUNABLE PROPERTIES**

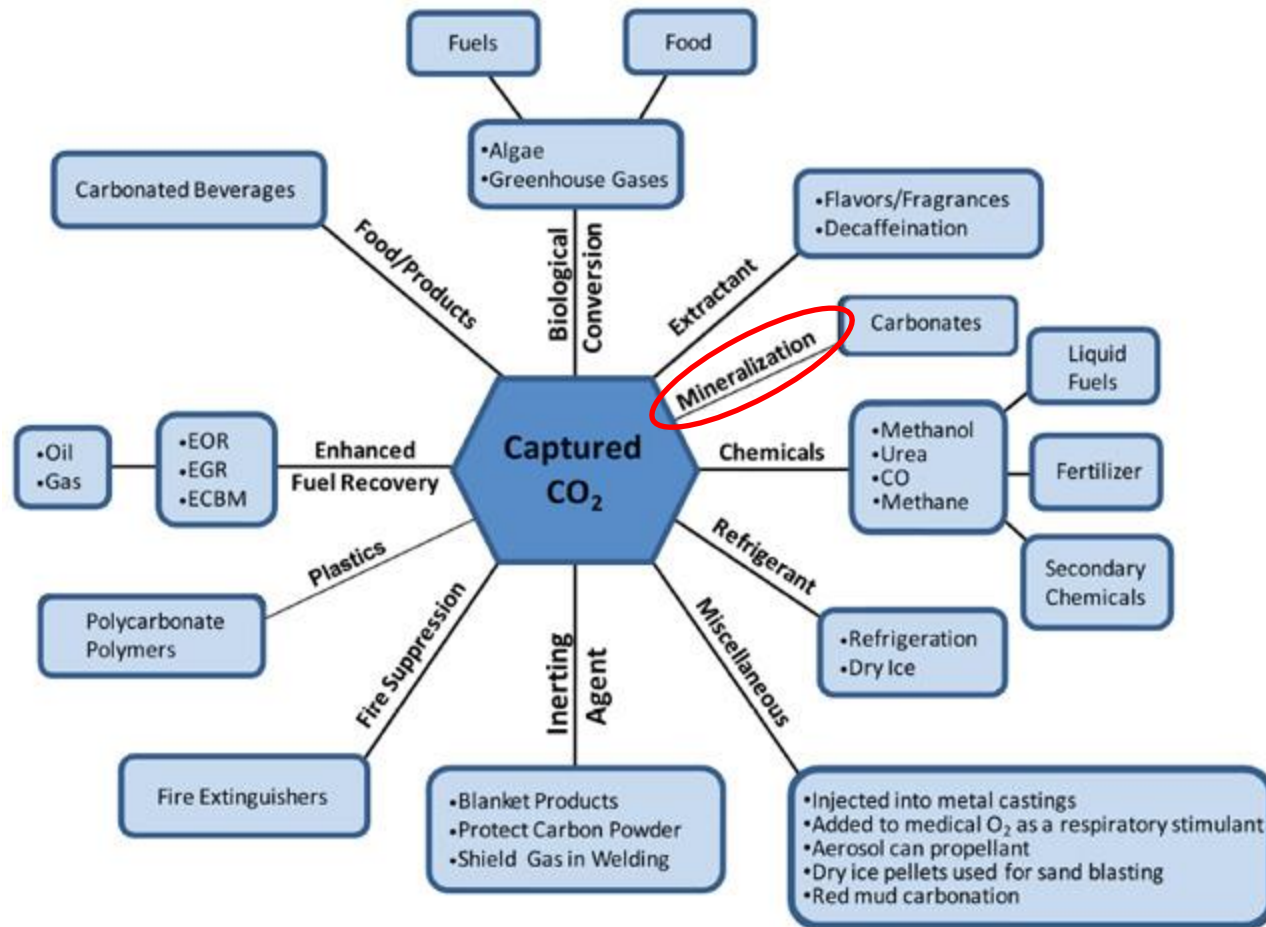
# Viscosity

## Effect of core size

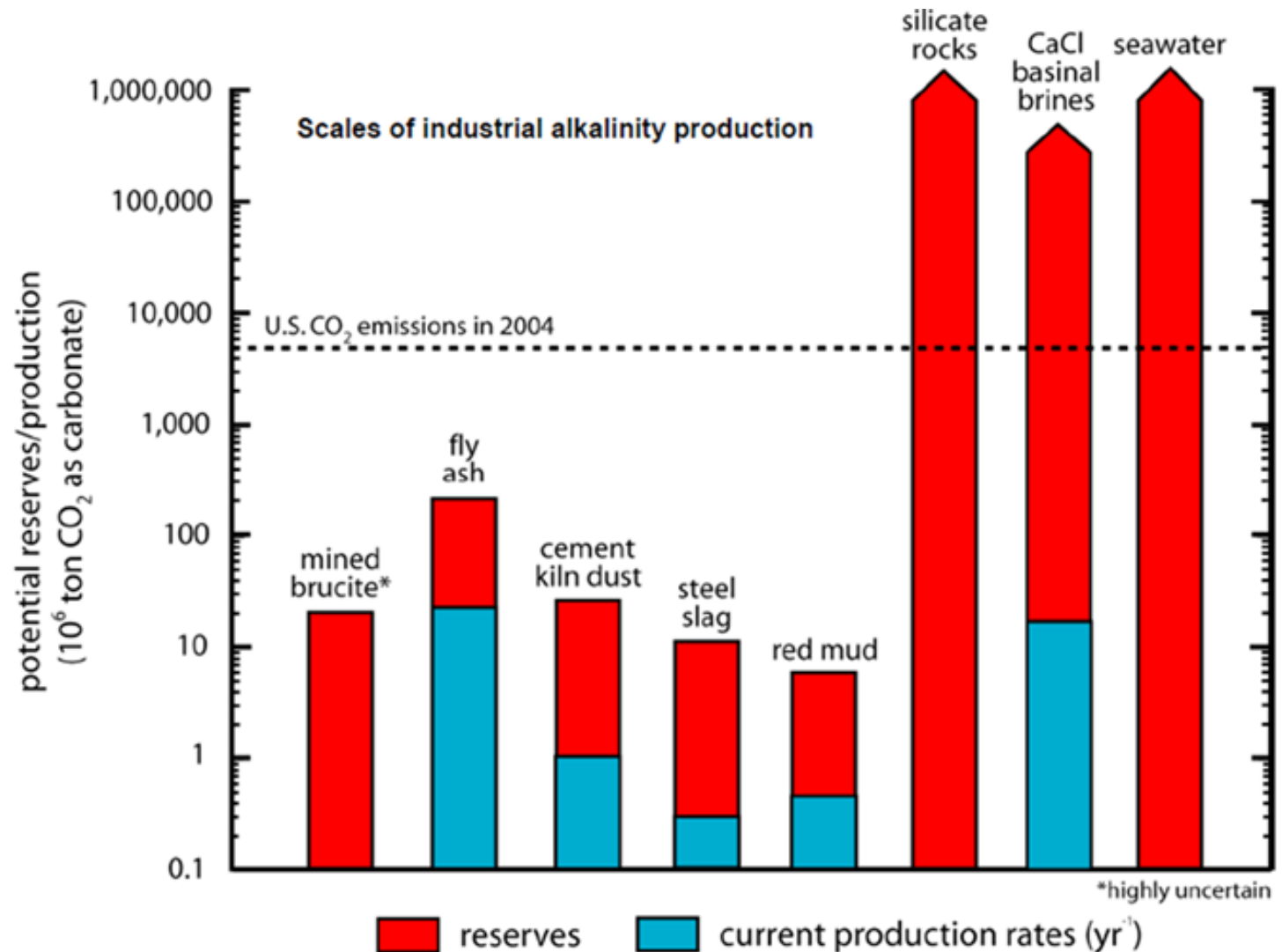


- Introduction of nanoparticles increases the viscosity of the system

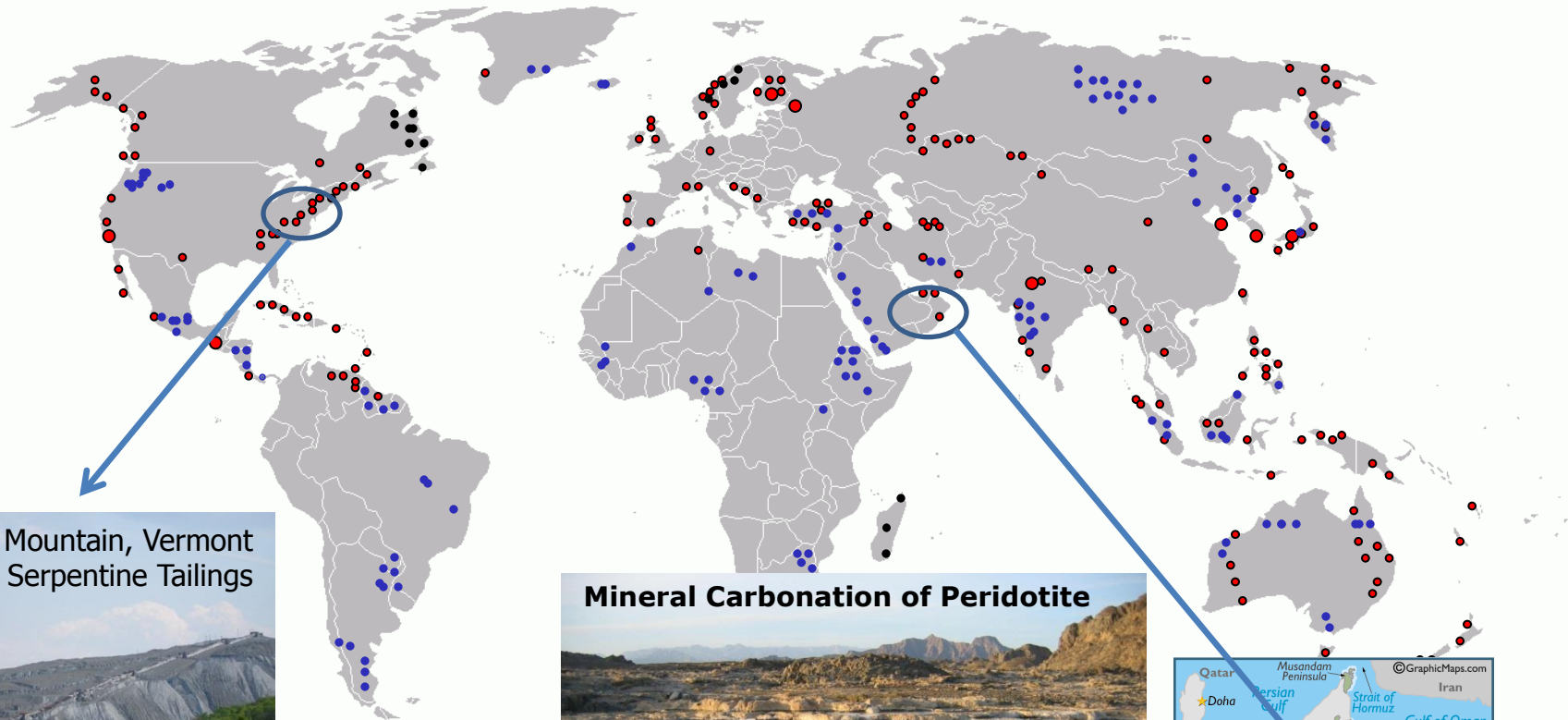
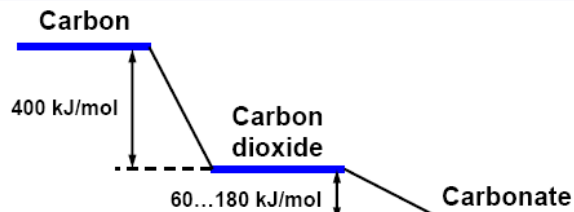
# CO<sub>2</sub> Utilization



# Carbonation of Industrial Wastes



# Carbon Mineralization



Belvidere Mountain, Vermont  
Serpentine Tailings



- Basalt
- Labradorite
- Magnesium-based Ultramafic Rocks (Serpentine, Olivine)

## Mineral Carbonation of Peridotite

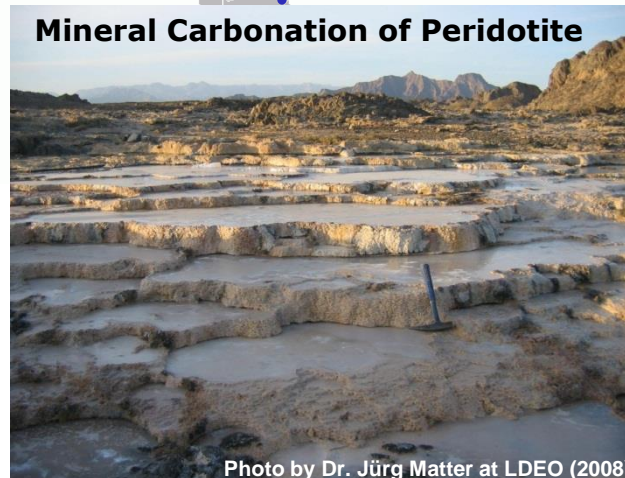
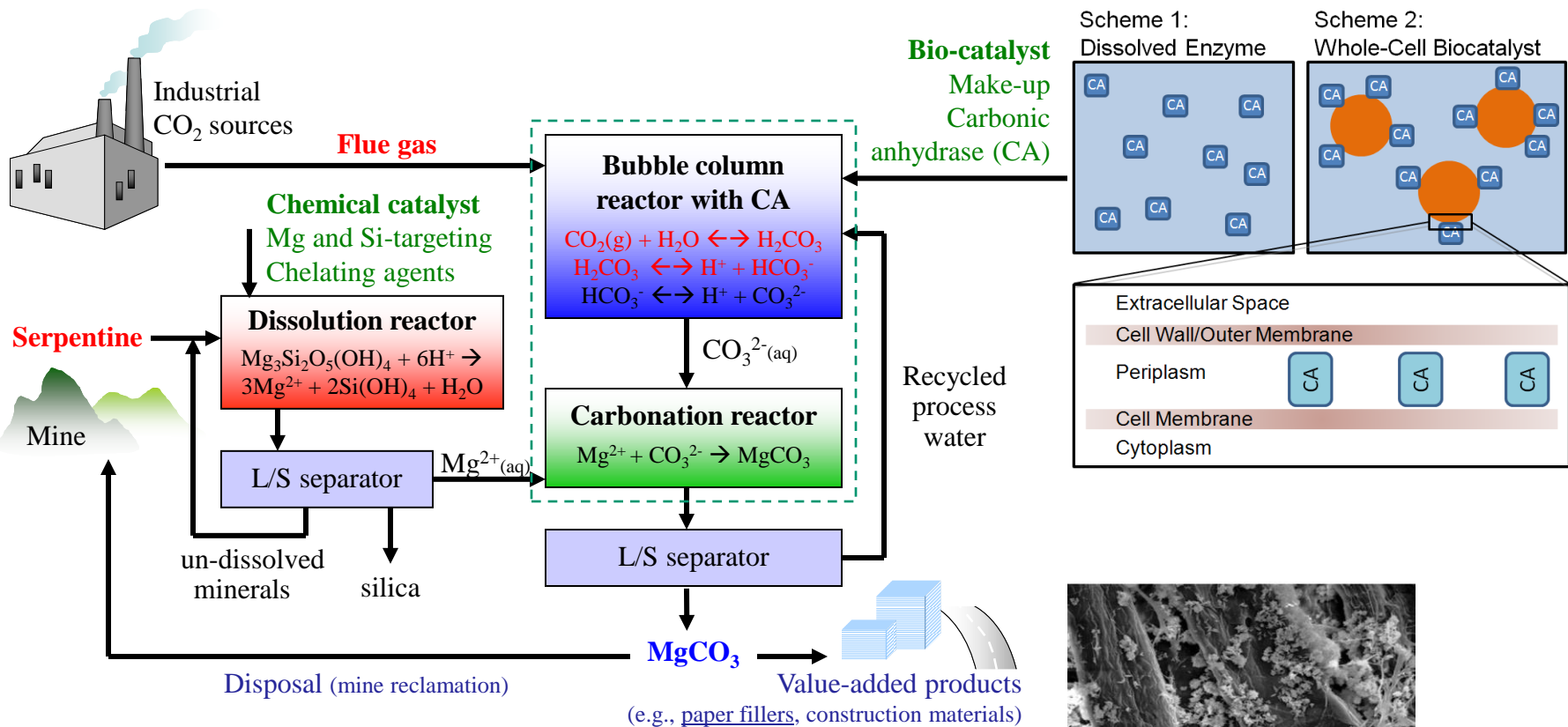


Photo by Dr. Jürg Matter at LDEO (2008)





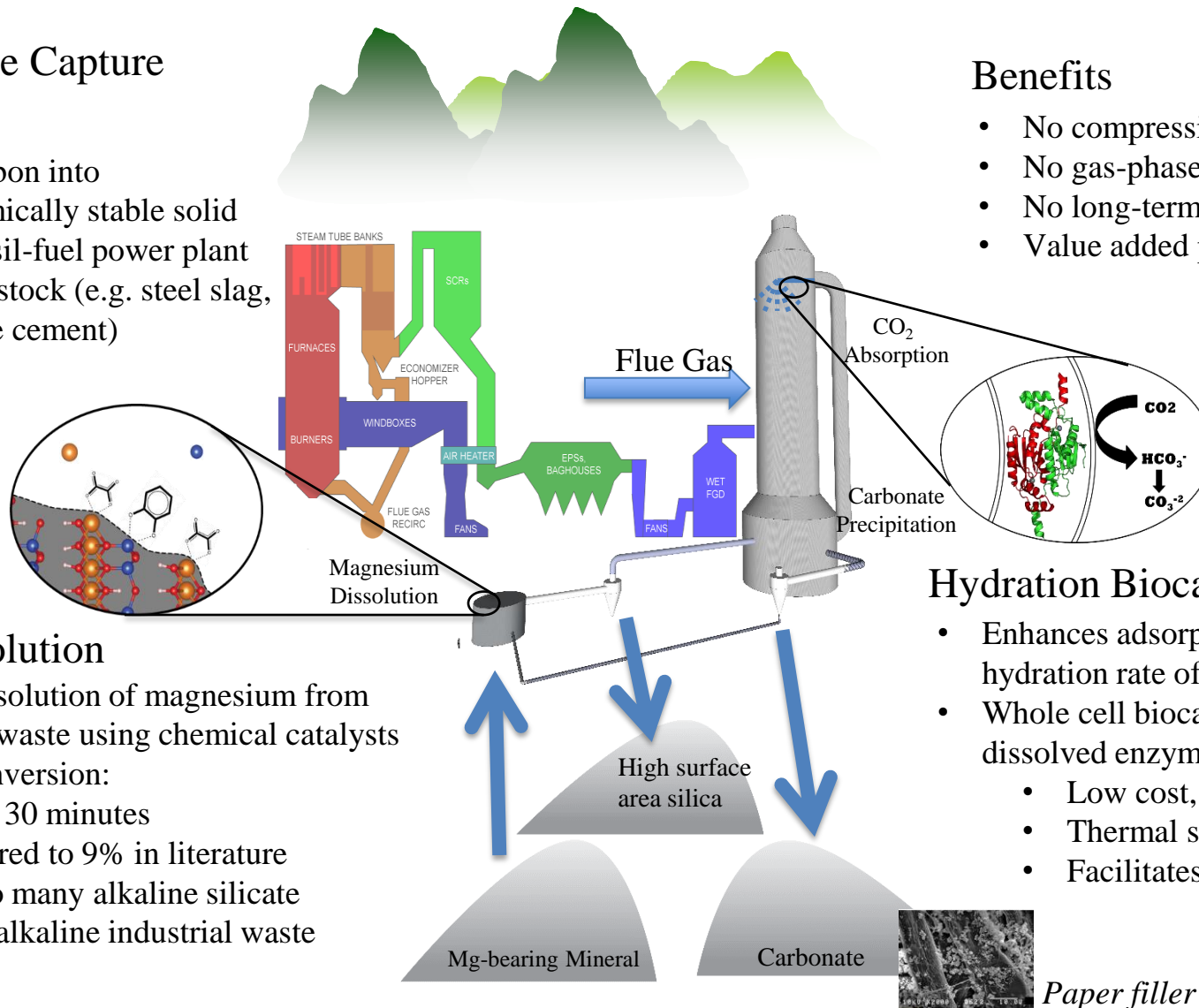
# Chemical and Biological Catalytic Enhancement of Weathering of Silicate Minerals as Novel Carbon Capture and Storage Technology



# Chemical and Biological Catalytic Enhancement of Weathering of Silicate Minerals as Novel Carbon Capture and Storage Technology

## Why Combine Capture and Storage?

- Converts carbon into thermodynamically stable solid
- Onsite at fossil-fuel power plant
- Flexible feedstock (e.g. steel slag, fly ash, waste cement)



## Benefits

- No compression
- No gas-phase storage
- No long-term monitoring
- Value added products possible

## Mineral Dissolution

- Enhanced dissolution of magnesium from minerals and waste using chemical catalysts
- Enhanced conversion:
  - 85% in 30 minutes
  - Compared to 9% in literature
- Amendable to many alkaline silicate minerals and alkaline industrial waste

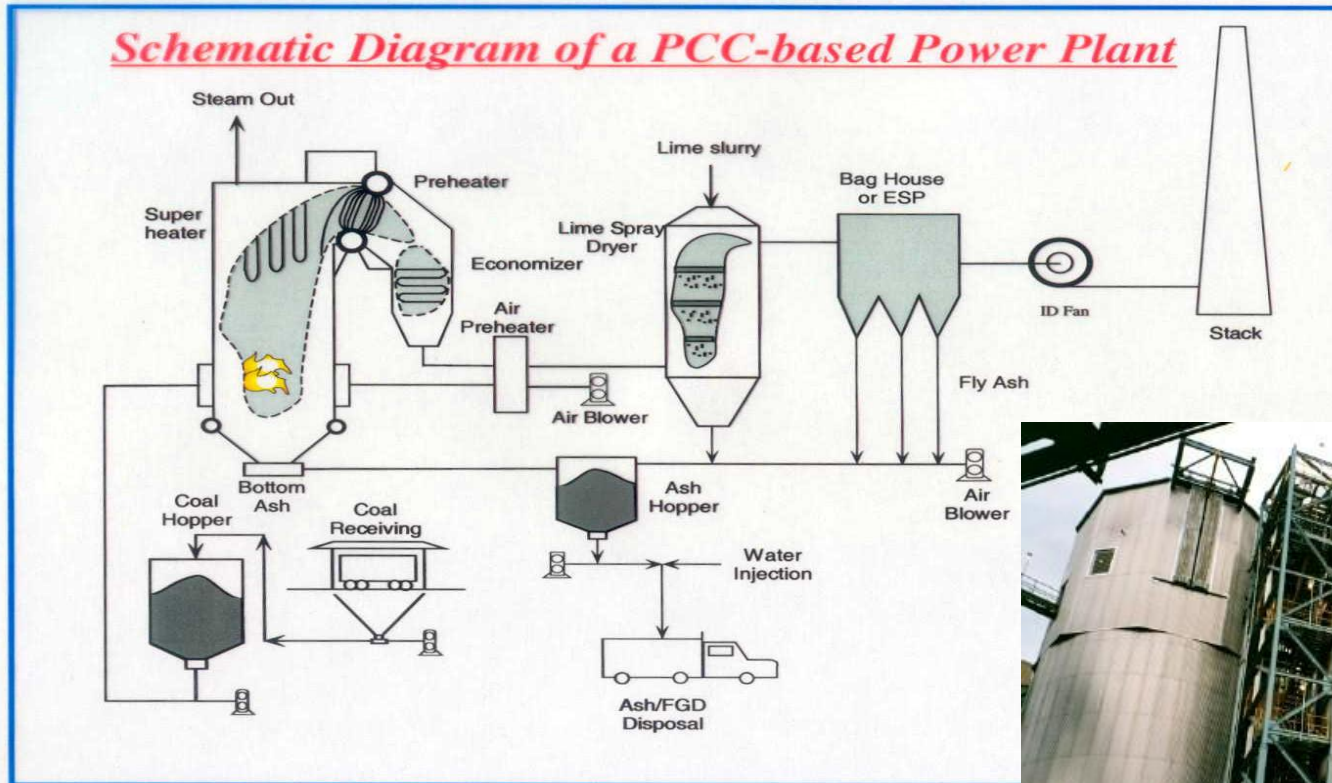
## Hydration Biocatalyst

- Enhances adsorption and hydration rate of carbon dioxide
- Whole cell biocatalyst vs. dissolved enzyme
  - Low cost, no purification
  - Thermal stability
  - Facilitates separation



*Paper filler*

# Scale of the Problem



The Ohio State University

**Combustion System – Lime Spray Dryer**



<http://www.xcelenergy.com>

# Current Challenges? Opportunities?

- Lower parasitic energy consumption (heat integration)
- Reduce the compression cost
- Viscosity issues for anhydrous solvents
- Moisture effect
- Multi-pollutant control
- Water requirement?
- Combined CO<sub>2</sub> capture and conversion
- etc

